

NOVAK, Jaroslav, inz.

Possibilities of increasing the productivity of rotary kilns
for magnetic roasting of sideritic ores. Sbor Vyzk ust Mnisek
4:41-52 '64.

Complete automation of rotary kiln processes. Ibid.:53-62

1. Research Institute of the Zelenorudne doly a hrudkovny
National Enterprise, Mnisek.

NOVAK, Jaroslav, muz.

Briquetting of iron ores, Kralupy nad Vltavou, Czechoslovakia

Research Institute, Zálesné Kralupy a Kralovky, Malesice
pod Brly.

NOVAK, Jaroslav, inz.

Technological samples for ore dressing research. Rudy 9 no.11:
375-376 N '61.

1. Vyzkumny ustav Zeleznych dolu a hrudkoven, Ejpovice.

(Ore dressing)

NOVAK, Jaroslav, inz.

Methods of low intensity magnetic separation. Rudy 9 no.11:377-379
N '61.

1. Vyzkumny ustav Zeleznych dolu a hrudkoven, Ejpovice.

(Iron) (Magnetic separation of ores)

"APPROVED FOR RELEASE: 07/19/2001

CIA-RDP86-00513R001137410011-2

NOVAK, J., ^{clerk} inz.

Magnetic noncontact recorder for computation of parts from
ferromagnetic materials. Automatizace 6 no. 6:150-151 Je '63.

APPROVED FOR RELEASE: 07/19/2001

CIA-RDP86-00513R001137410011-2"

NOVAK, Jaroslav

Heat capacity and the equation of state of solids. Pts.1-3.
Sbor VŠKMT Pardubice no.1:61-102 '64.

1. Vychodoceske chemicke zavody Synthesia National Enterprise,
Semin u Pardubic. Submitted June 18, 1963.

NOVAK,J

SOBK, F.; NOVAK, J.; HEROUT, V.

On terpenes. Part 51. The composition of chamaulene; preliminary communication [in English with summary in Russian]. Sbor.Chekh.khim. rab. 18 no.4:527-529 Ag '53. (MLRA 7:6)

1. Department of Natural Products, Institute of Organic Chemistry,
Czechoslovak Academy of Science, Prague.
(Chamaulene)

Kevork, J.

SCHL, F.; LIVSH, J.; KERK, A.

"Terpenes. Pt. 51. The Composition of Chamaemeles; A Preliminary Survey. II. n.
p. 15/7. (Chemical Listy. Vol. 47, No. 1, July 1953, Prague.)

See Lentz, List of East European Compounds, 2nd Ed., Part I, 1951.

CZECH

Tepner, LXIII. Total synthesis of chamazulene. A simple general synthesis of 1,4,7-substituted azulenes. J. Novák, František Šorm, and Jiří Sicher (Czech. Akad. věd, Praha, Československ. Chem. Listy 48, 1648-5X (1954); Collection Czechoslov. Chem. Commun. 19, 1264-73 (1954) [in English]; C. A. 49, 9084e.—A general synthesis of 1,4,7-trialkylazulenes has been worked out based on the hydrogenation of a suitably substituted ar-dihydroxybenzosuberene to the satd. diol, oxidation of the diol to the corresponding dicarboxylic acid, and cyclization to the desired azulene skeleton. Chamazulene was synthesized in this way. 2,3-(MeO)₂C₆H₃COCl and Pr₂Cd gave 72% 2,3-(MeO)₂C₆H₃COPr (I), b.p. 104°; semicarbazone, m. 154.5° (from MeOH). Refluxing 120 g. I, 61 g. Me₂NH.HCl, 35 g. (CH₂O)_n, and 2.5 ml. concd. HCl in 450 ml. EtOH 4 hrs. with efficient stirring, adding 10 g. (CH₂O)_n, refluxing 8 more hrs., distg. off most of the EtOH, stirring the cryst. residue several times with Et₂O, dissolving the crystals in H₂O, liberating the bases with 400 ml. 10% NaOH, extg. the mixt. with Et₂O, concn. the ext. *in vacuo* to 80 ml., and treating the concentrate with dry HCl gave 86 g. 2,3-(MeO)₂C₆H₃CO-CH₂(CH₂NMe₂)₂COHCl (II), m. 141° (from EtOH), and 86 g. recovered I. Treating 21.6 g. II in 60 ml. H₂O with 40 ml. 10% NaOH, extg. the liberated base with Et₂O, adding with cooling 20 g. MeI, evapg. the excess MeI and Et₂O *in vacuo*, stirring the crystals with 100 ml. EtOH, treating the mixt. with NaCH(CD₃)₂Et₂ [prepd. from 2.2 g. Na and 15 g. CH₂CO₂Et₂] in 100 ml. EtOH, refluxing the mixt. until no more Me₂N escaped (8 hrs.), distg. off the EtOH, dissolving the NaI in H₂O, refluxing the crude duster 2,3-(MeO)₂C₆H₃COCH₂CH₂CH₂CO₂Et₂ with 10 ml. 10% NaOH 2 hrs., acidifying the mixt., decarboxylating the free acid by heating 30 min. at 170°, and distg. the crude acid *in vacuo* gave 13.6 g. 2,3-(MeO)₂C₆H₃COCH₂CH₂CO₂Et₂ (III), b.p. 190°. Hydrogenation of 23.0 g. III in 900 ml. AcOH over 1 g. 5% Pd/C at 80-90° gave 30.0 g. 2,3-(MeO)₂C₆H₃CH₂CH₂CO₂Et₂ (IV), b.p. 185°. Cyclizing 23 g. IV by heating 1 hr. at 200° with 750 ml. phen

elien ✓ Kovář, J., and Novák, J.: Preparative Reactions in
Organic Chemistry. III. Nitration, Nitrosation, and
Sulfonation. Prague: Nakl. Českosl. akad. věd. 1956.
672 pp. Reviewed in *Chem. fürt 50*, 1860(1956).
Bibliography on Ethylene Imine Chemistry. Washing-
ton: Chemerad Corp. 1956. 65 pp. \$2.

NU. AK, S.R.A.

G-2

CZECHOSLOVAKIA/Organic Chemistry - Synthetic Organic Chemistry.

Abs Jour: Referat Zhur-Khimiya, No 5, 1958, 14358.

Author : Novak Jiri, Ratusky Josef, Sneberk Vladimir, Sorm Frantisek.

Inst :
Title : Reactions of Diazoketones. I. Interaction of Diazoacetone
with Unsaturated Compounds.Orig Pub: Chem. listy, 1957, 51, No 3, 479-492; Sb. chekhosl. khim.
rabot, 1957, 22, No 6, 1836-1851.

Abstract: Diazoketones [diazoacetone (I)] react with unsaturated compounds to form derivatives of acetyl cyclopropane. Reaction with aromatic compounds RH results in ketones $\text{RCH}_2\text{COCH}_3$ (II); C_6H_6 reacts only in the presence of BF_3 ; the yield of phenylacetone is 4%. Vinyl esters and ethers usually yield a mixture of cyclopropane derivatives and ketones II. Acrylonitrile forms with I (1 hour, 100° . addition of Cu) a compound

Card : 1/5

G-2

... organic Chemistry.

CZECHOSLOVAKIA/Organic Chemistry - Synthetic Organic Chemistry. G-2

Aba Jour: Referat Zhur-Khimiya, No 5, 1958, 14358.

only the acetate of 7-acetyl-bicyclo-[4,1,0]-heptanol-1, yield 12%, BP 72-76°/0.3 mm; SC, MP 184°. In the same manner from dihydropyran was obtained only 7-acetyl-2-oxabicyclo-[4,1,0]-heptane (IV), yield 54%, BP 80-83°/10 mm; MP 144°. IV was obtained also from the acid chloride of 2-oxabicyclo-[4,1,0]-heptane-7-carboxylic acid, MP 89-93°/15 mm (V--acid) and $(\text{CH}_3)_2\text{C}_4$ in C_6H_6 , yield 68%. V was obtained from dihydropyran and $\text{N}_2\text{C}(\text{CH}_3)\text{COOC}_2\text{H}_5$ with subsequent saponification with NaOH, yield 86%, MP 104°. Ipdole reacts with I in cyclohexane with addition of Cu at 80°; by distillation of the mixture and chromatographic purification is isolated, with a yield of 25%, 3-indolyl-acetone (MP 116.5-118.5°; picrate, MP 116-117°), which is reduced with LiAlH₄ to 1-(3-indolyl)-propanol-2, yield 89%, BP 146-147°/0.35 mm, MP 37.5°. Coumarone and CuSO₄ give with I at 80° 6-acetyl-2-oxa-3,4-benzobicyclo-[3,1,0]-hexene-3, MP 63°, which is converted with NaOBr in aqueous dioxane to 2-oxa-3,4-benzobicyclo-

Card : 4/5

CZECHOSLOVAKIA / Organic Chemistry. Synthetic Organic G-2
Novak, J.

Abs Jour: Ref Zhur-Khimiya, No 23, 1958, 77633.

Author : Novak, J. and Sorm, F.

Inst : Not given.

Title : Reactions of Diazoketones. III. The Reaction of
Diazoacetone with Furan and With Its Homologs.
A New Simple Synthesis of Octa-3,5-diene-2,7-dione.

Orig Pub: Chem Listy, 51, No 9, 1693-1698 (1957) (in Czech).

Abstract: In the course of their investigation of the feasibility of the utilization of diazoacetone (I) in the synthesis of other compounds, the authors have reacted I with furan (II) and with its homologs using 1 : 1 mol ratios. The reaction is accompanied by the evolution of N₂ and the opening of the ring of II. With II, 2, 4-heptadiene-6-one-al

Card 1/4

Nov. 20 1976
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Rec'd. 11/20/76, Prague

CZECHOSLOVAKIA / Organic Chemistry. Synthetic Organic G-2
Chemistry.

Abs Jour: Ref Zhur-Khimiya, No 23, 1958, 77633.

Abstract: (III) is formed, while sylvan (IIa) gives 3,5-octadiene-2,7-dione (IV) and 2,5-dimethylfuran (IIb) forms 4-methyl-3,5-octadiene-2,7-dione (V). The reaction apparently proceeds via the intermediate formation of 2-oxa-1,3,7-dicyclo-3-hexene. I and II are heated with Cu-powder in an autoclave (15 min, 90°); III is separated from the fraction boiling at 70-90°/0.4mm, yield 43%, mp 42°; the same fraction yields bis-dinitrophenylhydrazone, mp 207° (from CH₃ NO₂); bis-semicarbazone, mp 187°. When III is refluxed for 3 hrs with an aqueous solution of K₃ Fe(CN)₆ and CH₃ COOK, 1,3-hexadiene-5-one-1-carboxylic acid (VI) is obtained, yield 52%, mp 145° (from water), which under the action

Card 2/4

CZECHOSLOVAKIA / Organic Chemistry. Synthetic Organic G-2
Chemistry.

Abs Jour: Ref Zhur-Khimiya, No 23, 1958, 77633.

Abstract: of NaBrO in the cold gives trans, trans-muconic acid (VII), yield 70%, mp 303° (from water). VII is also obtained by the direct oxidation of III with NaBrO. The Hydrogenation of III over PtO₂ in alcohol gives 6-heptanone-1-al, bp 98-102°/15mm; hydrogenation over Raney Ni (120 atm) gives 1,6-heptanediol, bp 80°/0.3mm. The reaction of I with IIa is carried out by adding the mixture of reactants to a boiling suspension of Cu in IIa; the yield of IV is 39%, mp 127° (from CH₃OH). V is obtained by a similar procedure from I and IIb, yield 38%, bp 93-96°/0.3mm, mp 38°. The ethyl ester of diazoacetic acid under similar conditions

Card 3/4

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CZECHOSLOVAKIA / Organic Chemistry. Synthetic Organic G-2
Chemistry.

Abs Jour: Ref Zhur-Khimika, No 23, 1958, 77633.

Abstract: likewise cleaves the furan ring of IIa to form
the ethyl ester of VI, bp 105-115°/12mm; semi-
carbazone mp 162° (from alc). For Communication
II see RZhKhim, 1958, 50339. -- J. Kovar.

Card 4/4

Novak, J.

CZECHOSLOVAKIA/Organic Chemistry. Organic Synthesis. G-2

Abs Jour : Ref Zhur-Khimiya, No 9, 1959, 31331

Author : Novak, J., Sorm, F.
Inst : -

Title : Reactions of Diazoketones. III. Interaction of Diazacetone with Furan and Its Homologues. New Simple Synthesis of Octa-3,5-Dienedione-2,7.

Orig Pub : Collect. czechosl. chem. commun., 1958,
23, No 6, 1126-1132

Abstract : See Ref Zhur-Khimiya, 1958, 77633.

Card : 1/1

138

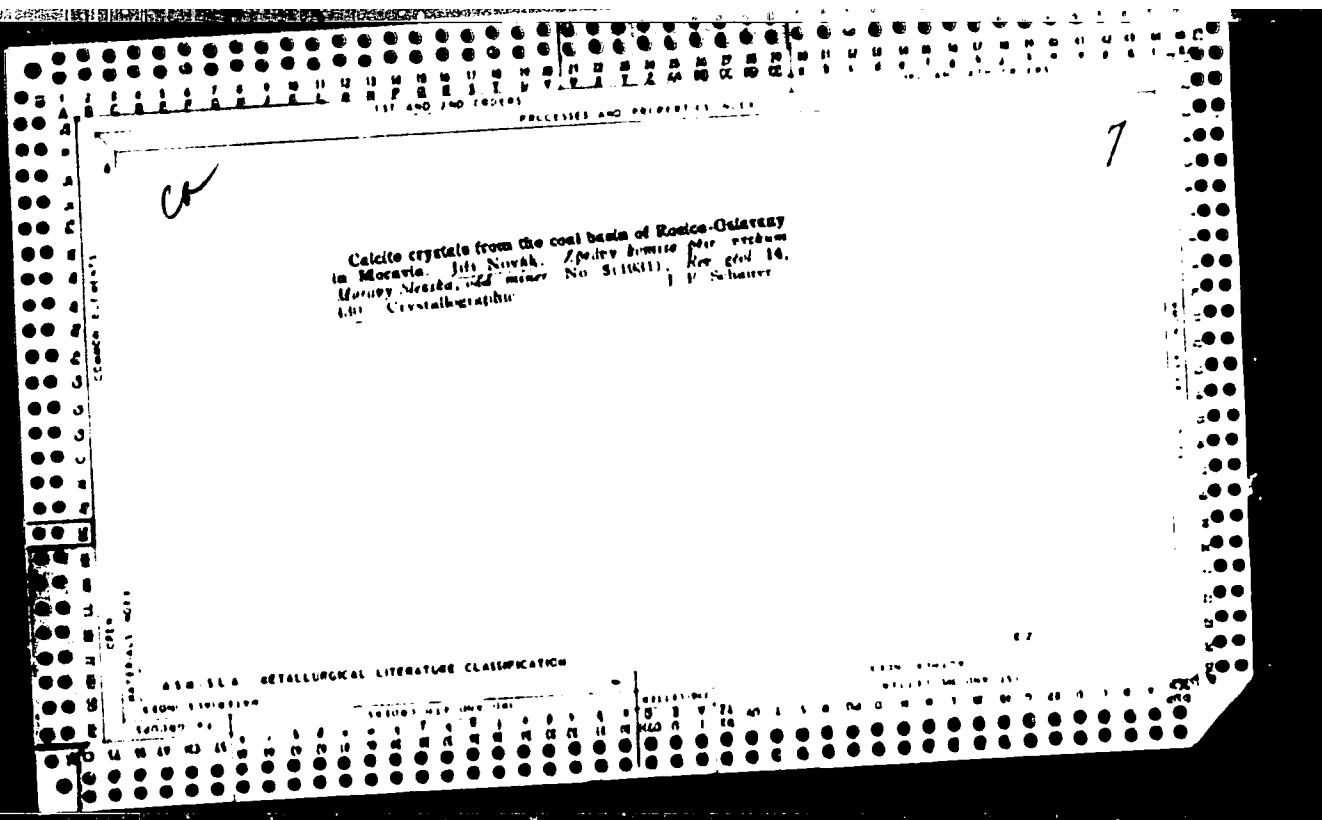
ADAMEK, Milan; NOVAK, Jiri

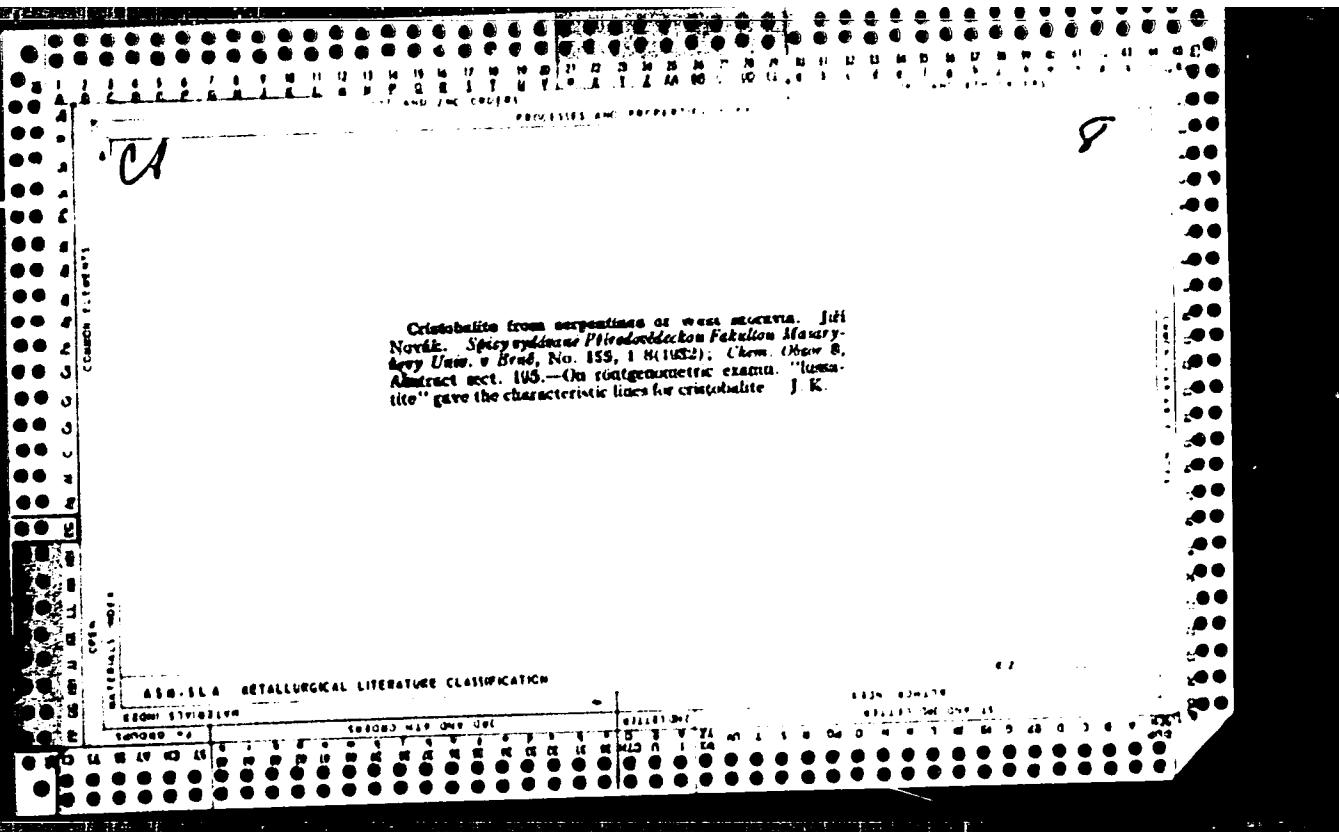
New linear polyesters on the basis of diphenyl sulfone-4,4'-dicarboxylic acid. Sbor VSChT Pardubice no.1:97-112 '63.

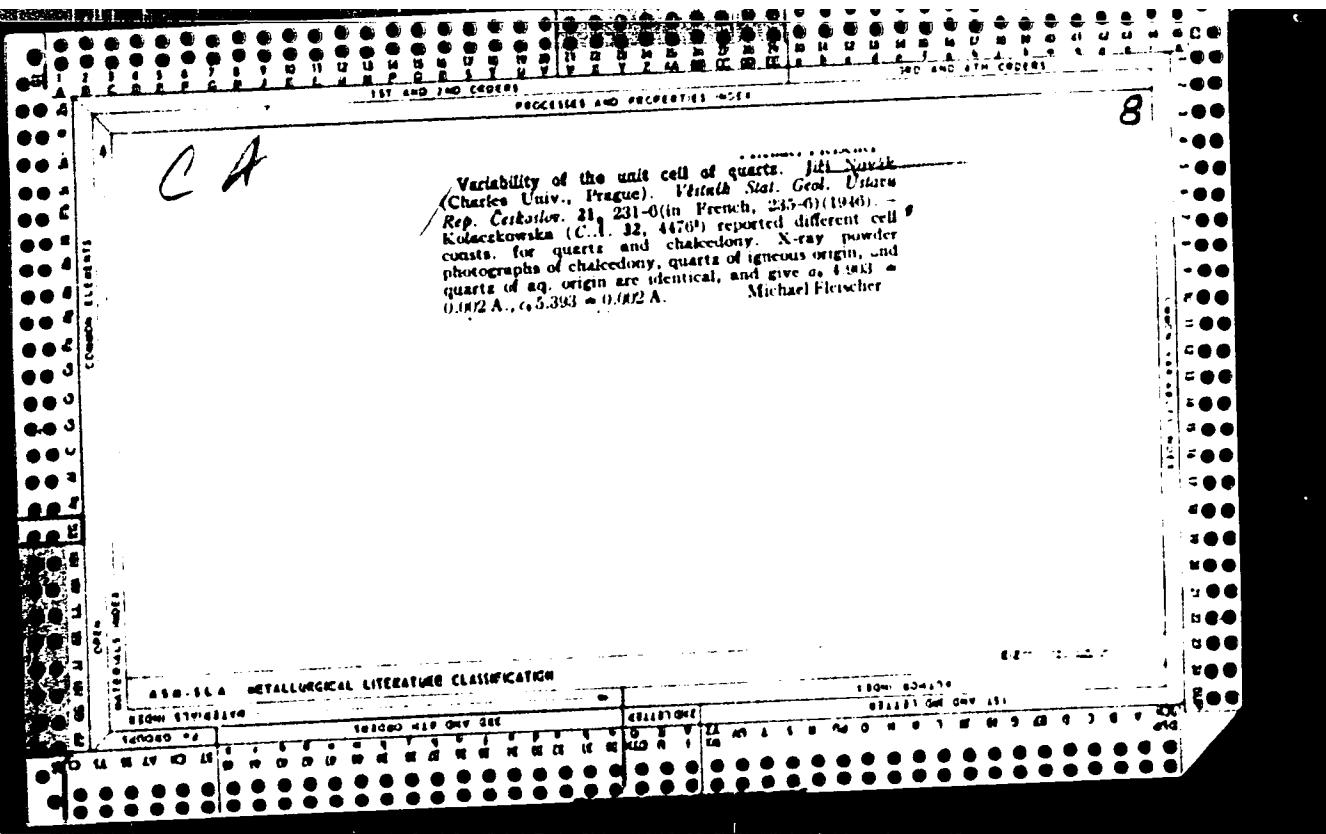
1. Chair of Plastics, Higher School of Chemical Technology, Pardubice.

NOVAK, Jiri

Contribution to the new chemical orthography and the question of
spelling foreign words. Chem listy 59 no.3:355-356 M. '65.







117. ON THE VARIABILITY OF THE CRYSTAL LATTICE OF QUARTZ.
—J. Novak (*Coll. Czech. Chem. Comm.*, 12, 193, 1847). Contrary to the experimental results of Koleckowka, who found different values for the lattice of chalcedony, of quartz of hydrothermal origin, and of quartz from a granite, the present author ascertained by the method of powder photographs that mixtures of quartz specimens of various origins all give the same lattice, viz. $a_0 = 4.903 \pm 0.002 \text{ \AA}$; $c_0 = 8.383 \pm 0.002 \text{ \AA}$.

Charles Ernest Phipps

NOVAK, JIRI.

Strukturní krystalografie; uvod do studia atomové stavby
hmoty. [1. vyd.] Praha, Prirodovedecke vydavatelstvi, 1953.
208 p. [Structural crystallography; an introduction to the
study of the atomic structure of matter. 1st ed. illus., bibl.,
index, notes, tables]

SOURCE: East European Accessions List (EEAL) Library of Congress.
Vol. 5, No. 1, January, 1956

NOVAK, J.

"Occurrence of Magnetite in the Krkonose (Riesengebirge) Mountains." p.5.
(Rudy, Vol.1, No.1, Feb. 1953, Praha.)

SO: Monthly List of East European Accessions, Vol.3, No.3, Library of Congress, March 1954,
Uncl.

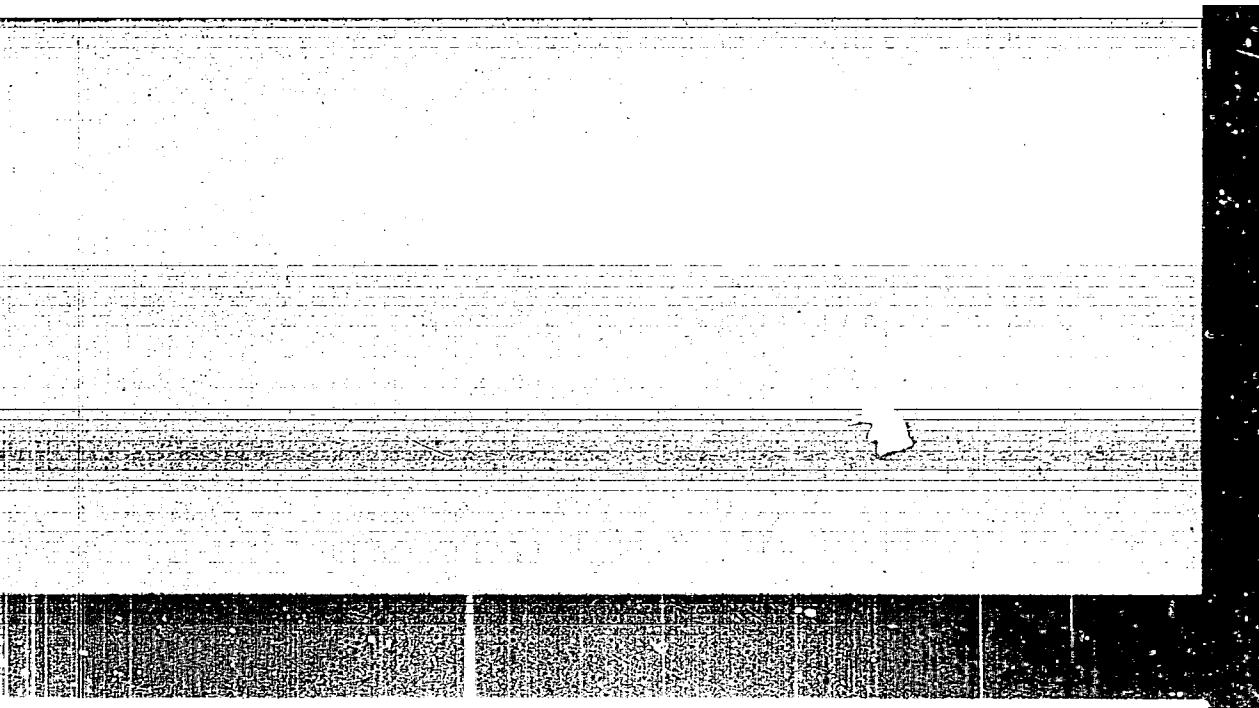
NOVAK, J.

"Metallurgy of the Caslav Area in Bohemia." p 121.
(Rudy, Vol.1, No.8, Oct. 1953, Praha.)

See: Monthly List of Last Year's Acquisitions, Mineralogical Museum, Prague, 1953.

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CIA-RDP86-00513R001137410011-2



APPROVED FOR RELEASE: 07/19/2001

CIA-RDP86-00513R001137410011-2"

NOVAK,

AUTHORS: Lisek, A. and Novak, J. CZECH/5-58-6-27/30
 TITLE: Meeting of the Czechoslovakian National Committee of the International Union of Crystallography (2. Porady) - Mezinárodní konference sestávající roční Krystalografické ústředny

PERIODICAL: Československý Casopis Pro Přírodu, 1958, Nr 6,
 pp 745 - 746 (Czech)

ABSTRACT: The Presidium of the second section to establish a National Committee with Prof. Dr. J. Novák as chairman, Dr. I. Kápar, Dr. J. Hanák, Candidate of Chemical Sciences, Prof. Dr. I. Šebek, Dr. J. Korbaová, Doctor of Physical-mathematical Sciences, Prof. Dr. I. Šebek, Corresponding Member of the Czech Ac. Sc., as member. At its meeting held on March 14, 1958, under the chairmanship of Prof. Dr. J. Novák, it was decided to co-operate in the publication of the "Handbuch für Minerogen und Gesteine" which is to be published as a joint effort of geologists from the USSR, East Germany, Poland and Czechoslovakia and is to contain practical information on structural analysis and to give in the form of tables and, possibly, graphs all the data required in practice for solving most problems of structural crystallography.

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Dr. V. Dynáček (Institute of Technical Physics of the Czech Ac. Sc.) was entrusted with this work. The National Committee has recommended participation in the efforts of Prof. R. Pepinsky, (Pennsylvania State University) in the publishing of an encyclopedic containing the structural, physical, chemical and other information of all substances identified. Collection of such data is of fundamental importance from the point of view of solving problems of physics of the solid phase, mineralogy, chemistry and other branches of science. For this purpose, it was proposed to form in Czechoslovakia a working team of crystallographers who could offer Prof. Pepinsky co-operation in the following fields: Processing of information published in the Czechoslovakian language; morphological and physical crystallographic measurements; supplementary X-ray data; calculation of the projections of the electron densities. As regards participation of Czechoslovakian crystallographers at the Fifth session of the International Crystallographic Union to be held in London in 1960, it is recommended that Czechoslovakian

work should be presented in the form of a symposium or that a foreign-language issue of one of the journals of the Czechoslovakian Ac. Sc. should be devoted to this purpose.

ASSOCIATION: Ištav technické fakulty USAF, Praha (Institute of Technical Physics of the Czechoslovakian Ac. Sc., Prague)
 Katedra geologico-minerologickej fakulty FJFI Praha
 (Chair of the Geological-Mineralogical Faculty of Charles University, Prague)

SUBMITTED: May 2, 1958

Card 3/3

Given Name

_____, Given Name

Country: Czechoslovakia

Academic Degrees: /not given/

Affiliation: /not given/

Source: Prague, Casopis pro Mineralogii a Geologii, Vol VI, No 2, 1961,
pp 221-222.

Date: "Fifth International Crystallographic Congress in Cambridge,
England."

Authors: NOVAK, Jiri

BOUSKA, Vladimir

SLANSKY, Ervin

GPO 96164

SOURCE: Given Names

Country: Czechoslovakia

Academic Degrees: /not given/

Affiliation: /not given/

Source: Prague, Casopis pro Mineralogii a Geologii, Vol VI, No 3, 1961,
pp 333-335.

Data: "Properties of Gossite and Its First Find in Nature."

Authors:

NOVAK, Jiri
ROST, Rudolf

GPO 98164)

NOVAK, Jiri; TBLIC, K., Monok

Discovery of deformed pyrite crystals in the Mtnik deposit,
near Hnusta. Cas min geol 9 no. 1:103-104 '64.

1. Prirodovedecka fakulta Karlovy univerzity; Ustav nerostnych
surovin, Kutna Hora.

KOVAK, Jiri K.

Obzor' otnoshenij na affinitet elementov k oxygenu i
sulfuru. Cas. mineral. 9, no. 1, 1963, p. 164.

1. Prirodovedeskij fakultet Karlovy univerzity.

L 42524-66

ACC NR: AP6005490

SOURCE CODE: 03/0078/00/000/001/301/0012

47

3

INVENTOR: Novak, Jiri (engineer) (image)

ORG: none

TITLE: [Electrometric circuit with a nonlinear dielectric element] (image) (cont'd.)
7120-64, Clas. Date:

SOURCE: Vynalezy, no. 1, 1966, 12

TOPIC TAGS: electrometer, electrometry, dielectric capacitor, phase detector

ABSTRACT: An electrometric circuit is described which has a nonlinear dielectric filter to the input clamps which consists of a high frequency generator whose output is connected to a capacitor with a nonlinear dielectric element. The distinguishing feature of this circuit is that it is designed to superpose a high frequency voltage of a proportional current on a duct capacitor with a nonlinear dielectric element with an ac voltage of the same frequency from a high frequency generator. A phase circuit and a differential rectifier for estimating the difference between the positive and negative amplitudes of the course of the superposed high frequency voltage constitute this design feature.

SUB CODE: 09/ SUBM DATE: 16Dec64

Card 1/1 2/2

NOVAK, Jiri, inz.

Vibrated large panels from CDK triple bricks. Poz stavby 11
no. 7:377-380 '63.

1. Pozemni stavby Pardubice.

NOV 2 1972

Experiments in automatic reading. Vodnik, Chap. 2, p. 228, 1940.

SCHUBERT, J.; NOVAK, J.

Conservative treatment of fetal erythroblastosis. Cas. lek. cesk.
91 no.3:72-77 18 Jan 52.

1. Gynekologicke, oddeleni st. obl. nemocnice v Motole.
Prednosta: Doc. dr. V. Sebek. Laborator pro stanoveni vlastnosit
krevnich pri internim oddeleni St. obl. nemocnice v Motole.
Prednosta: Prof. dr. V. Jonas.

(ERYTHROBLASTOSIS, FETAL, therapy
conservative, indic.)

NOVAK, Josef, MUDr.

Lumbo-sciatic syndrom and accidents. Prakt. lek., Praha 34 no.22:
502-507 20 Nov 54.

1. Z chirurg. klin. zakladny Ustavu pro doskolovali lekaru pri obv.
nemocnice v Praze 8 - Bulovka; prednosta prof. Dr. Jan Knobloch
(WOUNDS AND INJURIES

trauma in etiol. of lumbo-sciatic synd.)
(BACKACHE

lumbo-sciatic synd. caused by trauma)

(SCIATICA

lumbo-sciatic synd. caused by trauma)

NOVAK, Josef, MUDr

Surgical intervention in hemophiliacs. Rozhl.chir. 34 no.9:570-572
Nov 55.

1. Z chirurgicke klinické zakladny Ustavu pro doskoleni lekaru
pri obvodni nemocnici v Praze 8-Bulovka (Prednosta: profesor MUDr
Jan Knobloch)
(HEMOPHILIA, surgery,
(Cz))

NOVAK, Josef

Prognosis for ileus caused by gallstones. Cas. lek. cesk. 95
no. 46:1281-1284 16 Nov 56.

1. Chirurgicka klinicka zakladna Ustavu pro doskoleni lekaru
pri obv. nemocnici v Praze 8-Bulovka. Prednosta: profesor MUDr.
Jan Knobloch.

(CHOLELITHIASIS, compl.

intestinal obstruct., surg. progn. (Cz))

(INTESTINAL OBSTRUCTION, etiol. & pathogen.

cholelithiasis, surg. progn. (Cz))

NOVAK, Josef

Antibiotics in the treatment of appendiceal infiltrates. Rozhl.
chir. 38 no. 7:468-474 July 59

1. Chirurgicka klinicka zakladna Ustava pro dosklovaní lekaru v
Praze 8 - Bulovka, prednosta prof. dr. Jan Knobloch.
(ANTIBIOTICS, ther.) (APPENDIX, dis.)

NOVAK,J.; LOMSKY,J.; KUBAT,B.

Our experience with the diagnosis of acute gastrointestinal hemorrhage.
Cas. lek. cesk. 99 no.28:881-886 8 Jl '60.

1. Chirurgicka klinicka zakladna Ustavu pro doskoloovani lekaru v
Praze 8-Bulovka, prednosta prof. Dr. Sc. MUDr. Jan Knobloch. Chirurgicke
oddeleni OUNZ Cesky Brod, primar MUDr. J. Lomsky.
(HEMORRHAGE GASTROINTESTINAL diag)

Dr. J. K. H. G. VAN DER HORST, MD, PhD
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Chairman of Faculty of General Medicine at Amsterdam University
Attelierske en Antonius Postav faculty voor de medische leerdienst van Amsterdam
Amsterdam, Holland (present address) 1007 ZW, The Hague, Holland

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Amsterdam, Holland (present address) 1007 ZW, The Hague, Holland

NOVAK, Josef

TECHNOLOGY

Periodical AUTOMOBILE. Vol. 2, no. 12, 1958.

NOVAK, J. Developmental trends in the electric equipment of motor vehicles. p. 124.

Monthly List of East European Accessions (EAI) M, Vol. 8, no. 3, March, 1956. Incl.

SOUSTEK, Milos; NOVAK, Josef

Experience with using pallets in railway transportation services. Zel dop tech 10 no.4:112-113 '62.

NOVAK, Josef

Accessories of motor vehicles. Tech praca 14 no.6:485-490
Je 62.

1. Prisluzenstvi motorovych vozidel, sdruzeni narodnich podniku,
Praha.

KLIKA, Otakar, prof, inz.; NOVAK, Josef, inz.

A small linear analyzer for solving transport problems. Slaboproudý
obzor 23 no.9:500-504 S '62.

1. Ceske vysoke ucení technicke, Praha.

NOVAK, Josef, inz., C.Sc.

Increasing technical and organizational standards in handling
materials up to 1970. Tech praca 15 no.4:249-253 Al '63.

1. Statni komise pro rozvoj a koordinaci vedy a techniky,
Praha; predsedu komise pro manipulaci s materialem pri
Ustredni rade Ceskoslovenske vedecko-technicke spolecnosti.

NOVAK, Josef, inz. CSc.

Handling of materials is the main source of reserves up to
1970. Tech praca 17 no.2:81-83 F '65.

1. State Commission for the Development and Coordination of
Science and Technology, Prague.

CZECHOSLOVAKIA / Analytical Chemistry. Analysis of
Inorganic Substances.

E-2

Abs Jour: Ref Zhur-Khimiya, 1958, No 17, 57187.

Author : Dolezhal J., Novak J.^{ref}

Inst : Not given.

Title : Use of Aminocompounds in the Polarography of In-
organic Substances. VI. Determination of Cobalt
in Steel and Ores.

Orig Pub: Chem. listy, 1957, 51, No 10, 1798-1803.

Abstract: Based on the results of polarographic behaviour
of Co in the alkaline solutions of ethylenediamine,
a determination method for Co in the presence of
considerable quantities of Fe, Al, Cr, Mn, Ni, Zn,
V, Mo, and W has been developed. Large quantities
of Ag, Tl and Cu interfere with the determination
since the reduction proceeds at higher potentials

Card 1/5

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CZECHOSLOVAKIA / Analytical Chemistry. Analysis of
Inorganic Substances.

E-2

Abs Jour: Ref Zhur-Khimiya, 1958, No 17, 57187.

Abstract: than that of Co (-0.7 volts in a solution, 1 M basis ethylenediamine and 1 M basis KOH; -56 volts in a solution, 1 M basis ethylenediamine and 1 M basis NH₄OH). Pb also interferes since its wave merges with that of Co. In the above stated media and at high concentrations of Fe, Cr, Mn, Al, Zn, and V, it forms precipitates which adsorb the complex of Co(3+). With the decreased concentration of ethylenediamine, the adsorption increases, but with the increased concentration of Co and at equal concentration of Fe, it decreases. By employing a method of standard dosages added, it is

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CZECHOSLOVAKIA / Analytical Chemistry. Analysis of
Inorganic Substances.

E-2

Abs Jour: Ref Zhur-Khimika, 1958, No 17, 57187.

Abstract: possible to determine Co in the presence of Fe for the ratios of Co: (Fe, Mn, Zn) = 1:1000, Co : Ni = 1 : 100, Co : V = 1 : 10. In the determination of Co in steel, a 0.1gr sample is dissolved in 7cc of 20% H₂SO₄ or any other suitable acid. The obtained solution is then reduced in volume by a factor of 2 by means of heating with 4cc of concentrated H₂SO₄. After cooling, 3cc of 20% H₂SO₄ is added to the dry residue, followed by the addition of the KOH solution until pH of the total solution becomes approximately 2.0. The solution is then diluted to 100cc. 12.5cc of freshly prepared ethylenediamine (2 M basis ethylenediamine, and 2 M basis KOH) is added to a 10cc portion of the above solution. The mixture is then thoroughly mixed,

Card 3/5

18

CZECHOSLOVAKIA / Analytical Chemistry. Analysis of
Inorganic Substances.

E-2

Abs Jour: Ref Zhur-Khimika, 1958, No 17, 57187.

Abstract: heated to a required temperature, diluted with water to 25cc, centrifuged (or retained standing until the precipitate settles), and the transparent solution is subjected to polarographic determination after the removal of oxygen by means of bubbling with N₂. In the determination of Co from raw minerals, 1gr sample is dissolved in 10cc aqua-regia, evaporated to dryness, followed by two evaporation using 10cc of concentrated HCl each time. 10cc of concentrated HCl is then added to the dry residue and, after reaching a required pH, the solution containing suspended undissolved material,

Card 4/5

CZECHOSLOVAKIA / Analytical Chemistry. Analysis of E-2
Inorganic Substances.

Abs Jour: Ref Zhur-Khimya, 1958, No 17, 57187.

Abstract: is diluted to 100cc with water, followed by the performance of a test as described above. If the analyzed sample contains large quantities of Pb and Cu, they should be removed beforehand. For Part V refer to Ref Zhur-Khimya, 1958, 32166.

Card 5/5

19

100-141-10
CZECHOSLOVAKIA / Analytical Chemistry. Analysis of Organic Substances.

Abs Jour: Ref Zhur-Khimiya, 1958, No 20, 67335

Author : Janak J., Novak J.

Inst : Not given.

Title : Chromatographic Semi-micro Analysis of Gases. XIV.
Direct Determination of Individual Gaseous Paraffines and Olephines in 1, 3-Butadiene.

Orig Pub: Chem. listy, 1957, 51, No 10, 1832-1837.

Abstract: Chromatographic method capable of handling mixture of liquid and vapor hydrocarbons was developed for determining contaminants present in commercial butadiene (I). This method employs apparatus in

Card 1/3

. CZECHOSLOVAKIA / Analytical Chemistry. Analysis of Organic Substances.

Abs Jour: Ref Zhur-Khimija, 1958, No 20, 67335.

Abstract: the form described previously (Ref. Zhur-Khimija, 1954, 18572) with the following modifications: ahead of the chromatographic column an additional column is installed. It is filled with kieselguhr impregnated with maleic anhydride. Butadiene is adsorbed in this column at a temperature of 100-110°. Contaminants that dissolve partially in maleic anhydride are subjected to the preliminary separation. The final separation occurs at room temperature in the main column which is filled with "aluzyt" (sodium aluminum silicate) impregnated with 20% dimethylformamide. CO₂ is used as carrier gas. Chromatographical spectra of the gaseous hydrocarbons are obtained from the main column effluents. These gases constitute contamin-

Card 2/3

CZECHOSLOVAKIA / Analytical Chemistry. Analysis of
Organic Substances.

Abs Jour: Ref Zhur-Khimiya, 1958, No 20, 67335

Abstract: ants of commercial butadiene and consists of ethylene, propane, propylene, isobutane, n-butane n- and iso-butlenes, trans-butylene, cis-butylene, iso-pentane, and n-pentane. The obtained spectra not only detect the presence of each component, but also determine its quantity thus enabling to determine purity of I, which is also analyzed in the same apparatus. Each determination requires 5-50cc of gas. Duration of each determination is approx. 30 minutes. Error encountered in the determinations of each individual component is less than 0.02% (absolute). For Part XIII refer to Ref. Zhur-Khimiya, 1958, 32195.

Card 3/3

CZECHOSLOVAKIA/Analytic Chemistry. Analysis of Inorganic
Substances.

E

Abs Jour: Ref Zhur-Khim., No 23, 1958, 77339.

Author : Dolezal, Jan; Novak, Josef.

Inst :

Title : Rapid Methods of Analysis of Metals and Mineral
Raw Materials. V. Polarographic Determination of
Copper and Bismuth in Mineral Raw Materials and
Iron.

Orig Pub: Chem. listy, 1958, 52, No 1, 36-39.

Abstract: Copper can be determined in the presence of Fe^{3+} after
the reduction of Fe^{3+} with hypophosphorous acid.
The reduction of Fe^{3+} in the medium of 8 M HCl pro-
ceeds very slowly even at low concentrations of Fe^{3+} ,
and in the medium of 2 M HCl, it proceeds consider-

Card : 1/4

90

CZECHOSLOVAKIA/Analytic Chemistry. Analysis of Inorganic Substances.

E

Abs Jour: Ref Zhur-Khim., No 23, 1958, 77339.

ably more rapidly even in the case of higher Fe concentrations. In the presence of 10^{-4} M Hg (NO_3)₂, Fe^{3+} in 0.5 M FeCl_3 is reduced after 3 min. of boiling of the solution. 5. 10^{-4} M of Cu in the presence of a 10,000-fold amount of Fe can be determined by the second wave of Cu ($\text{Cu}^{\text{t}} \rightarrow \text{Cu}^{\text{i}}$) prepared separately after the reduction in the medium of 2 M of HCl and 0.1 M of NaH_2PO_4 . Bi and Pb can be also simultaneously determined, if their concentrations are approximately the same as that of Cu. Mo and Ti interfere with the determination of Cu, but As does not. The described method was applied to the

Card : 2/4

CZECHOSLOVAKIA/Analytic Chemistry. Analysis of Inorganic Substances.

E

Abs Jour: Ref Zhur-Khim., No 23, 1958, 77339.

determination of Cu in ores. 1.5 g of a finely pulverized sample is dissolved in 5 ml of concentrated HCl and 1 ml of concentrated HNO₃; it is evaporated twice with 2 ml of concentrated HCl, heated 10 min. with 20 ml of water and 2 ml of concentrated HCl in a water bath, cooled, and the solution together with the precipitate is diluted with water to 100 ml. 2 aliquot portions (10 ml each) of the prepared solution are poured into measuring flasks of 25 ml capacity, and a standard addition of Cu is added into one of them. After that, 5 mg of 10 n. HCl, 3 ml of 5 M NaH₂PO₄, and 0.1 ml of 0.1 n. Hg(NO₃)₂ are added into each flask, the flasks are boiled 3 min. until the solu-

Card : 3/4

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CZECHOSLOVAKIA/Analytic Chemistry. Analysis of Inorganic
Substances.

E

Abs Jour: Ref Zhur-Khim., No 23, 1958, 77339.

tion becomes colorless, they are cooled, 2.5 ml
of 1% gelatin solution is added to each, the
contents are diluted to the mark and, after having
the solutions bubbled with nitrogen, they are
polarographed. See RZhKhim, 1958, 57166 for
report IV. - Petr Zurn.

Card : 4/4

CZECHOSLOVAKIA/Physical Chemistry. Electrochemistry.

D

Abs Jour: Ref Zhur-Khim., No 5, 1959, 14797.

Author : Dolezel J., Novak J.

Inst :

Title : Polnographic Behavior of Trivalent and Tetraivalent Cerium.

Orig Pub: Chem. listy, 1958, 52, No 4, 582-588.

Abstract: A clear reversible reducing-oxidizing wave of the $\text{Ce}(4+)+e \rightleftharpoons \text{Ce}(3+)$ system is observed in solutions of alkaline carbonates of Ce, at pH 9 - 11.5. At carbonate concentrations < 0.75 M at pH < 8.7 or at pH > 12.7 , the wave height decreases and a precipitate is formed. In 1.0 M K_2CO_3 $E_{\frac{1}{2}}^{\circ} = -0.114$ volts and in 2.0 M K_2CO_3 $E_{\frac{1}{2}}^{\circ} = -0.158$ volts (satur. K.F.).

Card : 1/2

Distr: 4E2c

7

Use of amines in inorganic polarography. VII. Polarographic determination of copper and iron in nickel and aluminum and their salts. Jan Doleček and Josef Novák (Karlová Univ. Prague). Česk. 1952, 1953-1958, cf. C.A. 42, 20446. Simultaneous detn. of Co and Fe was possible in a supporting electrolyte consisting of 0.8M ethylenediaminetetraacetic acid and 0.08M K₂P₂O₇. In this soln. the waves of Fe and Cu are well septd. F. Štrížek

7

44 84

NOVAK

AUTHORS: Dvořák, J. CZECH/9-52-11-020
 TITLE: Rapid Analytical Method for Metals and Inorganic Raw Materials. XVI. Polarographic Determination of Cerium Alloys and Inorganic Matter (VI). (Soviet journal of analytical chem. (Zhurn. Vsesoyuznogo Nauchno-Issledovatel'skogo Instituta po Analiticheskym Metodam), Vol. 52, No. 11, pp. 2000 - 2005)

PERIODICAL: (Czechoslovakia), 1978, Vol. 52, No. 11, pp. 2000 - 2005

ABSTRACT: The continuously increasing demand for cerium (Ce) as a component of the most diverse alloys has led to the investigation of the problem of rapid and direct determination in a wide range of concentrations as well as in trace amounts, side by side with a whole series of other elements in the analyzed material. The growth in the uses of Ce is also linked with new sources of raw material and its testing requires more rapid and sufficiently accurate analytical laboratory methods.

Up to now, Ce has been determined by measurement (Ref. 1) by weight, Ref. 2 by colorimetric methods (Ref. 3) or polarographic methods have hardly ever been used because the reduction of the Ce ions in the proposed electrolytes becomes apparent either as an increase in current from the zero line of the Galvanometer (Ref. 4-6) which is unsatisfactory for accurate measurements, or causes difficulties of reading rather negative potentials (Refs. 4, 7). This is normally preceded by depolarization of a whole series of Ce elements which then interfere with its polarographic determinations (Ref. 8).

One of the previous communications (Ref. 8) gives a detailed account of the polarographic behavior of Ce or Ce⁴⁺ valence in alkaline carbonate solutions. The experience gained in this work was used for the rapid determination of Ce made a number of improvements and also research in different materials for alkaline, anhydrous carbonates, alloy, rotteneite after Ce electrolysis, in Ce containing steel and in zirconia sand.

The apparatus and reagents used for the analytic determination of Ce are the same as those described in the previous communication (Obz. Listy, 1978, Vol. 52, p. 2010). The chemicals used and their preparation were identical. The polarographic determination of Ce is carried out in solutions in which 1.1% precipitate after analytical decomposition in a 1% nitric acid. The main attention is given to the effect of oxidation. Ce determination firstly by absorption and also by titration. Reduction or oxidation gave (owing to the high solubility values of the products as carbonates).

In greater than 1 M K₂CO₃ solution is a linear function of the concentration within range 5.10⁻⁵ M to 10⁻⁴ M Ce³⁺. However, the majority of the ions are precipitated in this environment and thus affect Ce determination firstly by absorption and also by titration. The wave occurring in carbonate solutions in an inert atmosphere, the wave being due to carbonate complex [Ce(CO₃)₂]²⁻. Its amplitude is greater than 1 M K₂CO₃ solution is a linear function of the concentration within range 5.10⁻⁵ M to 10⁻⁴ M Ce³⁺.

The pure carbonate as electrolyte is the variable pH and carbonate concentration when the sample contains a fair amount of mineral acids. This would also affect the amplitude and shape of the wave (Ref. 9). Preliminary experiments were made with strongly acidic analytical solutions with carbonate buffer and also indicator dye. In all cases unacceptable errors were obtained and therefore these methods were rejected. A 0.5% solution of sodium citrate was experimentally tried and also failure. Other agents tried were pyrolytic citrate and tartarate. Other agents tried were sodium tartrate containing of 2 M K₂CO₃ and 0.1 - 1 M sodium-potassium tartrate gives a well measurable anodic wave of the Ce oxide of the carbonate complex while

Ce reduction gives a cathodic wave of diffuse character, its amplitude has a linear dependence on concentration and the square root of the height of the red wave. The half-wave anodic potential in 2 M carbonate and 1 M potassium-sodium tartrate is -0.162 V, of the cathodic -0.175 V (E.P.) and is almost the same as that obtained with pure

21. FeCO_3 . Polarographic analysis of the anodic curve points to an error of 5% or and of the cathodic of 6%.
 Although the literature mentions the possible formation of a complex between Ce³⁺ and the tartrate (Ref 2), the almost identical half-wave potentials in the mixture and the pure carbonate solution and also the logarithmic analysis of the curves suggest that only the carbonate ions are participating in the formation of the complex. Further experiments were made to test the effect of the concentration of other occurring elements and of rare earths on the anodic wave given by Ce in the 2 test solutions. The experiments were carried out so that to an acid solution of Ce³⁺ together with another element was added such an amount of tartrate that in a volume of 25 ml. its concentration was about 1.2 M; the carbonate solution was then added. All solutions were first freed of oxygen. This procedure must be followed very accurately because when the tartrate is added it has a marked effect as a buffering agent.
 The pH is then adjusted to optimal before the carbonate is added. The results of Ce determination aside by side with other elements are given in Table 2. The anodic waves of As, Hg, Pd, and Au precede that of Ce while the one for Mn is rather oblique and fuses with the Ce one. When present in larger amounts, these metals will affect the results. Cobalt ions also interfere with Ce determination and their presence cause wave deformation. The cathodic wave of Cu has about the same half-wave potential as the anodic Ce wave. Ferric and vanadyl ions in large concentration can also affect the Ce determination. In the proposed solution ferric ions will form a greenish yellow complex at 0.2 N Fe³⁺ concentration. Above this limit the half-wave potential for Ce will shift towards a positive value (by about 20 mV) and the limiting current cannot be easily subtracted. Nevertheless, up to 1.8 Ce in Fe can be determined with sufficient accuracy. The initially forming precipitate in the presence of Al (0.4 N Al³⁺) dissolves again when the carbonate is added and the solution is shaken. In some Al concentrations contributes to a more perfect shape of the anodic wave and to its easier observation. Using the standard addition method one can determine 2% Ce in all with the normal error of polarographic analysis.

Similar reactions (precipitation and its disappearance) occur with Zn (above 0.4 M of Zn^{2+}), Cr, Ti, Pb, Cd, Li, Mo and V. Va and Ni ions (0.2 M) form a precipitate but do not affect the results of Ce determination. Ca in Mn or Va can be determined when using the standard addition method. With the standard addition method one can also determine 0.1% Ce in Th and U, 0.5% Ce in La and Be. It appears that Cr, Nd, and Eu will not affect Ce determination but insufficient amounts of each of these rare elements were available to test the effect of higher concentrations. Pure carbonate solutions can be used in the case of Th and U (Ref 2) because they form carbonato complexes giving a better limiting current at the cathodic wave.

Several examples of analyses are then given of the Ce determination in Al³⁺-Ce³⁺-tartrate mixtures, in alloy and other Ce-containing materials such as barium titanate sand. There 1 figure, 5 tables and 9 references, 3 of which are Czech, 2 English, 2 German and 1 Italian.

ABSTRACT:
 Katedra analytiké chemie, Karlovo univerzity, Charles University,
 Prague,
 January 1, 1976

SUBMITTED:

Card 8/6

(3 /

Novak J.

Distr: 4E2c

Rapid analytical methods for metals and minerals. V.
Polarographic determination of copper and bismuth in
minerals and in iron. J. Dolčák and J. Novák. Collection
of Czechoslov. Chem. Commun., 24, 81-819. (In Ger-
man).—See C.A. 52, 19602b. M. Hudlický

7
Chromatographic semimicroanalysis of gases. XIV.
Direct determination of individual gaseous paraffins and olefins in 1,3-butadiene. J. Janák and J. Novák. Collection
Czechoslov. Chem. Commun. 24, 384-86 (1959) (in German).
See C.A. 52, 18604. XV. Automation of the measuring unit of a gas chromatograph. J. Janák and K. Tesářík.
ibid, 530-44 (in German). See C.A. 52, 20526.

L. J. Urbaňek /

Distr: 4E2c(j)

GW

41

5
2 May
1

Jed

COUNTRY	:	Czechoslovakia	8-17
CATEGORY	:		
ARG. J.R.R.	:	RZhKhim., No. 5 1960, No. 2-109	
AUTHOR	:	Dolezel, J. and Novak, J.	
INST.	:	Not given	
TITLE	:	On the Polarographic Behavior of Cerium(III) and Cerium(IV)	
CITE. PUB.	:	Collection Czechoslov Chem Commun, v4, no 7, 21-24 1959 (1960)	
ABSTRACT	:	See RZhKhim, 1960, No 5, 14797.	

CARD: 1/1

56

84/5b

Z/008/60/054/011/004/005
E112/E453

5.5600 (1273, 12P2 on 17)

AUTHORS: Novák, Josef, Rusek, Miroslav and Janák, Jaroslav

TITLE: Apparatus Using Flame-Ionization Detection

PERIODICAL: Chemické listy, 1960, Vol. 54, No. 11, pp. 1173-1182 + 1 plate

TEXT: Factors controlling the design of a Czechoslovak high-temperature gas chromatography apparatus using a flame ionization detector are discussed and operating data are given. A diagrammatic lay-out of the apparatus and of the electrical circuit in the detector are shown. A photograph of the complete and mass-produced instrument is included. Its design follows conventional lines. The apparatus is housed in a thermostat suitable for a temperature range up to 350°C and capable of employing four columns, each of them U-shaped and approximately 850 mm in length. Thermostating is achieved by hot air which is made to circulate by means of a fan. The design of the flame ionization detector differs from that introduced originally to gas chromatography by I.G. McWilliam (Gas Chromatography, ed. D.H. Desty, Butterworth Scientific Publications, London, 1958, p.142). In the McWilliam model the issuing gas is being burnt at a jet made from a hypodermic needle which, at the same time, serves as the positive pole for

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Z/008/60/054/011/004/005
E112/E453**Apparatus Using Flame-Ionization Detection**

the flame conductance measurements. The negative pole consists of a piece of 30 mesh brass gauze about 1 cm above the needle. In the Czechoslovak design the jet is a steel capillary of 0.5 mm bore above which are placed two platinum wire electrodes on top of each other and perpendicular to the axis of the flame. In series with the electrodes is a battery of 120 to 200 V and a resistor, across which a recorder is connected. The following operating data were investigated.

- a) Time factor for the establishing and maintenance of constant temperature parameters.
- b) Maintenance of constant pressure conditions at varying flow-rates of carrier gases. After the introduction of mercury manostats, the deviations of pressure at flow rate changes of 100% amounted to no more than 1.5%.
- c) Effect of voltage on signal sensitivity at constant weight of chromatographed compounds. Results are summarized in a table indicating voltages required for complete ionization at different flow rates and varying distances of electrodes.
- d) Effect of variations of N₂ and H₂ flow on signal sensitivity and stability of base line. Sensitivity is more affected by change in N₂-flow rate than that of H₂.

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Apparatus Using Flame-Ionization Detection

The effect of N₂-flow on one hand on the stability of the base line is negligible while that of H₂ is considerable

e) Effect of electrode distance and of their shape. Sensitivity decreases as the distance between the electrodes increases and a linear relationship is shown to exist. Position and shape of upper electrode is not of great importance but detector can function properly only on condition that the lower electrode is placed within the ionized space. The optimum distance of the lower electrode from the tip of the jet is determined by the maximum concentrations of the eluted fractions passing through the detector. Flame profiles and temperature contours are studied in this connection. Ionization gradients leading to inversions of chromatograms and measures for their elimination are discussed. Inversions are more likely to occur at low flow-rates

f) The performance of the instrument as a tool of analytical chemistry is discussed and results of analyses of different mixtures are given. A special study is devoted to partition chromatography homologues It is considered that the concentration of ions per mole in a series of homologous

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E112/E453**Apparatus Using Flame-Ionization Detection**

hydrocarbons increases in proportion to the number of carbon atoms in the molecule. The conductivity of the flame is explained by the formation of carbon ions, which are considered a transitory stage before the combustion of the hydrocarbons⁷ to carbon dioxide. In compounds, therefore, in which the carbon atoms may already be in an oxidation stage, a transition to carbon ions is not possible and these compounds will show characteristics associated with compounds having a lower number of carbon atoms e.g. acetone will behave like a compound containing only two carbon atoms. Compounds which do not contain carbon will show similar characteristics. The method has been found also very suitable for the analyses of biological mixtures and for medicinal work (acetone assay in diabetic urine etc). Other examples include results of analyses of coal tar constituents. Column packings, temperatures and other details are tabulated for each particular type of analyses. Illustrations that are included show the general arrangement of apparatus, a schematic diagram of the electrical circuit of the detector and a photograph of the complete instrument. There are

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Z/008/60/054/011/004/005
E112/E453

Apparatus Using Flame-Ionization Detection

13 figures, 7 tables and 15 references. 7 Czech and 8 English.

ASSOCIATION Laborator^y pro analysu plynů ČSAV Brno
(Laboratory for Gas Analysis Czechoslovak Academy
of Sciences Brno)

SUBMITTED: March 5, 1960

Card 5/5

NOVAK, Josef

"Simple Preparation of Definite Gaseous Mixtures for Calibration in Gas-Chromato-graphic Analysis of Trace Amounts of Compounds, Chemicke Listy, No. 11, Nov 60, Prague, p. 1189.

Affiliation: Lab for Gas Analysis, CSAV, Prague.

NOVAK, J.; FRIED, V.; PICK, J.

Solubleness of carbon dioxide in water under various pressures and temperatures. Coll Cz Chem 26 no.9:2266-2270 '61.

1. Institut fur physikalische Chemie, Technische Hochschule fur Chemie, Prag.

(Carbon dioxide)

NOVAK, J.

Determination of small amounts (up to 0,1 g/m³) aromatic hydrocarbons in the lighting gas and carbonization gases.
Paliva 41 no.3:84-87 Mr '61.

1. Ceskoslovenska akademie ved, laborator pro analyzu plynu,
Brno.

44862

S/081/62/000/024/022/073
B117/B186AUTHOR: Novák, J.

TITLE: Smoothing of pressure oscillations in the carrier gas in gas chromatography

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 24, 1962, 152,
abstract 24B1039 (Collect. Czechosl. Chem. Comms.,
v. 27, no. 2, 1962, 411-423 [Ger.; summary in Russ.])

TEXT: To stabilize the carrier gas pressure in gas chromatography either buffer capacity or pneumatic resistors such as capillaries have hitherto been used. It was shown theoretically that neither of these gave satisfactory results and that a combination of capacity and pneumatic resistors was necessary." Based on the similarity between pneumatic and electric systems and on determining the smoothing factor as the ratio of the relative pressure oscillation at the outlet and the relative pressure oscillation at the inlet, the theory of smoothing devices for pressure oscillations was considered. For this purpose a simple device in the form of a pneumatic three-stage resistor was suggested. The dependence of the

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S/081/62/000/024/022/073
B117/B186

Smoothing of pressure...

Pressure oscillation amplitude at the outlet on the oscillation frequency and amplitude at the inlet was studied in order to compare this device with a simple capillary. An example of a chromatogram of an illuminating-gas mixture showed that with the device suggested an ideal smoothing of the pressure oscillations, i. e., a smooth zero line can be obtained on the chromatogram. [Abstracter's note: Complete translation.]

Card 2/2

JANAK, J.; NOVAK, J.; SULOVSKY, J.

Separation of substituted malonic acid ester by gas-liquid chromatography and a new method of its identification. Coll
Cz Chem 27 no.11:2541-2549 N '62.

1. Laboratorium fur Gasanalyse, Tschechoslowakische Akademie
der Wissenschaften, Brno und Farmakon, Olomouc.

2

JANÁK, J; NOVÁK, J; ZÖLLNER, G.

Czechoslovakia

Laboratory for Gas Analysis, Czechoslovak Academy
of Sciences -- Brno - (for all-Zöllner presently in
Budapest, Hungary)

Prague, Collection of Czechoslovak Chemical Communi-
cations, No 11, 1962, pp 2628-2636

"Separation of Ethylamine in Presence of Ammoniac
and Water through Gas-Fluidity-Chromatography."

NOVAK, Josef; AREND, Hanus

Determination of titanium in the presence of a small quantity of iron
in barium titanate. Silikaty 7 no.2:150-154 '63.

1. Fyzikalni ustav, Ceskoslovenska akademie ved, Praha.

NOVAK, Josef

Analysis of the momentum, heat and mass transfer in turbulent flow systems with small transversal motion. Jaderna energie 9 no.8:265 Ag '63.

1. Statni vyzkumnny ustav tepelne techniky, Praha.

NOVAK, Josef; JANAK, Jaroslav

Operational properties of the prototype of high-temperature
gas chromatograph Chrom II. Chem listy 57 no.4:371-389
Ap '63.

1. Laborator pro analysu plymu, Ceskoslovenska akademie ved,
Brno.

ACCESSION NR: AP4016123

Z/0012/64/000/001/0059/0062

AUTHOR: Novak, Josef; Arend, Hanus

TITLE: Determination of fluorine in aluminum oxide, by means of melting and decomposition of the sample in a mixture of sodium carbonate, silicon and titanium oxides

SOURCE: Silikaty, no. 1, 1964, 59-62

TOPIC TAGS: fluorine colorimetry, aluminum nitride analysis, corundum analysis, fluorine determination, sodium carbonate, silicon, titanium oxide

ABSTRACT: A new method of colorimetric determination of fluorine is based on the decolorization of complexes of zirconyl ions with xylenol orange due to the presence of fluorine. It is suitable for the determination of F in single crystals of synthetic corundum, and in aluminum nitride. It is fast, accurate and well reproducible. The sample is melted in a Pt crucible with sodium carbonate, Ti and Si dioxides, digested with water, and the water solution filtered and added to the mixture of zirconyl nitrate and xylenol orange. Loss of color after 24 hours is compared to a set of standard colors. Orig. art. has 2 tables.

ASSOCIATION: Fysikalni ustav CSAV (Institute of Physics,

Card 1/1 Submitted: 8 Jy 63 Czechoslovak Academy of Sciences), Prague

NOVAK, Josef

Economic importance of the vaporization loss reduction.
Ropa a uhlie 6 no. 4: 120-123 Ap '64.

1. Ministry of Chemical Industry, Prague.

ACCESSION NR: AP4033425

Z/0055/64/014/004/0247/0255

AUTHOR: Sicha, M.; Vesely, V.; Novak, J.; Pekarek, L.

TITLE: Determination of the relaxation time of the electron temperature in the positive column of the electric discharge

SOURCE: Chechoslovatskiy fizicheskiy zhurnal, v. 14, no. 4, 1964, 247-255

TOPIC TAGS: relaxation time, electron temperature, electric discharge, electron density, positive column

ABSTRACT: A method of measuring the relaxation time of the temperature of electrons in the positive column of an electric discharge is described. The method uses measurements of the phase shift between the course of the electron temperature and that of the concentration of electrons in artificially excited moving striations of small amplitude. These data and the values measured for the electric field and temperature of the electrons in a homogeneous column are used to calculate the relaxation time of the electron temperature on the assumption that the diffusion of the electron temperature has no substantial influence on the time. The authors conclude that their results indicate that theoretical

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ACCESSION NR: AP4033425

mastery of the layer phenomena in the positive column plasma has already opened new possibilities in plasma diagnostics. Orig. art. has: 10 formulas and 3 tables.

ASSOCIATION: Lehrstuhl fur Elektronik und Vakuumphysik der Karlsuniversitaet, Prague(Chair of Electronics and Vacuum Physics, Charles University); Physikalisches Institut der Tschechosl. A.d.W., Prague(Physics Institute, Czech. Academy of Sciences)

SUBMITTED: 06Nov63

DATE ACQ: 01May64

ENCL: 00

SUB CODE: GP

NO REF Sov: 002

OTHER: 011

Card 2/2

L 61536-65 EWT(m)/EPF(n)-2/T/EPA(bb)-2 Pu-4
ACCESSION NR: AP5019184

CZ/0038/64/010/011/0407/0407

20
19
B

AUTHOR: Novak, Josef

TITLE: Dimensionless analysis and conditions of similarity of the temperature field in the hot piping and armatures of the primary cycle of the A-1 nuclear power plant in case of a dangerous temperature drop

SOURCE: Jaderna energie, v. 10, no. 11, 1964, 407

TOPIC TAGS: ¹⁹nuclear power plant, nuclear power technology, temperature characteristic/ A-1 nuclear power plant

Abstract: /author's English summary, modified/ Starting from the dimensionless solution of differential equations, the author specifies the conditions of simulating the temperature field in the hot piping and armatures of the primary cycle of the A-1 nuclear power plant. Five modifications of the piping and armatures are examined. From the conditions of simulation, directions are derived for designing suitable experimental models, and instructions are given on how to test these models. Brief data are given also on the assembly for testing the models. SVUTT Report 61-08006/1961.

"APPROVED FOR RELEASE: 07/19/2001

CIA-RDP86-00513R001137410011-2

L 61536-65

ACCESSION NR: AP5019184

ASSOCIATION: Statni vyzkumny ustav tepelne techniky, Prague (State Research
Institute of Heat Engineering)

SUBMITTED: 00

ENCL: 00

SUB CODE: NP, TD

NB REF NOV: 000

OTHER: 000

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Card 2/2

APPROVED FOR RELEASE: 07/19/2001

CIA-RDP86-00513R001137410011-2"

L 62734-65 EWP(t)/EWP(b) IJP(c) JD

ACCESSION NR: AP5021409

CZ/0034/64/000/012/0884/088621

17
B

AUTHOR: Novak, Josef; Panek, Zdenek

TITLE: Contribution to the determination of lead in copper alloys

SOURCE: Nutnicke listy, no. 12, 1964, 884-886

27,53

TOPIC TAGS: polarographic analysis, lead, electrolyte, copper alloy

Abstract: The article describes an improvement in polarographic determination of lead by the use of a new electrolyte solution. This solution is composed of 0.5 - 3.5% KCN, 5 - 25% NaOH, 0 - 8% Chelaton I (III), 0 - 2% of hydrazine sulfate, 0.001 - 0.2% of gelatin, and 0 - 5% sodium sulfite. This electrolyte is suitable at much higher concentration of Mn than previously used solution. The amount of both Fe and Mn may be up to 5% without interfering with the analysis. 0.005 to 5% of Pb may be determined. The results

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are more reliable than the colorimetric or electrolytic methods mainly in low contents of Pb. Orig. art. has: 2 tables, 2 figures, 1 graph.

ASSOCIATION: Novak - Fysikalni ustav CSAV, Prague (Institute of Physics, CSAV);
Panek - Vyzkumny ustav CKD, Prague (Research Institute CKD)

SUBMITTED: 00

ENCL: 00

SUB CODE: MM, GC

NR REF Sov: 001

OTHER: 019

JPRS

Card 2/2

NOVAK, Josef

Purification of chloroform and carbon chloride used as extraction agents in analytic chemistry. Chem listy 58 no.11:1338-1339 N '64.

1. Institute of Physics, Czechoslovak Academy of Sciences, Prague.

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1. V.A.C., Dated:

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L 31413-66

ACC NR: AP6022955

SOURCE CODE: CZ/0008/65/000/009/1021/1037
*K3
B*

AUTHOR: Novak, Josef

ORG: Laboratory for Gas Analysis, CSAV, Brno (Laborator pro analysu plynu CSAV)

TITLE: Quantitative analysis by means of gas chromatography / I. Some general relationships between the quantity of chromatographed compound and the size of the corresponding chromatographic curve

SOURCE: Chemicke listy, no. 9, 1965, 1021-1037

TOPIC TAGS: gas chromatography, viscosity, heat conductivity, physical diffusion

ABSTRACT: The determination of the concentration of a substance which is calculated on the basis of an area limited by a chromatographical curve, is correct only when the properties of the vapors of the analyzed substance and of the carrier gas are additive, and the concentration of the substance in the gas is low. When the determination is based on properties that are a function of the dynamic properties of the substance (viscosity, thermal conductivity, diffusion) a correction factor should be used. This equals: $1/(d\sigma_i/d x_i)$, where $\sigma_i = \sigma_{10} - \sigma_o$; σ_o is the investigated property of the carrier gas, and σ_{10} the corresponding property of the mixture; x_i is the concentration of the analyzed substance in the carrier gas. The author thanks Engineer, Doctor of Sciences J. Janak, Engineer, Candidate of Sciences M. Krejca, and Graduate Chemist K. Tesarik for valuable remarks on the manuscript of this work. Orig. art. has: 63 formulas. [JRS]

SUB CODE: 07, 20 / SUBM DATE: none / ORIG REF: 005 / SOV REF: 001 / OTH REF: 042
Card 1/1 JT

L 42263-66

ACC NR: AP6031474

SOURCE CODE: CZ/0008/66/000/003/0345/0346

17
B

AUTHOR: Novak, Josef

ORG: Institute for Physics, CSAV, Prague (Fysikalni ustav CSAV)

TITLE: Elimination of traces of iron from chemicals by the use of 1,10-phenanthroline without reducing agents

SOURCE: Chemicke listy, no. 3, 1966, 345-346

TOPIC TAGS: buffer solution, chemical purity

ABSTRACT: A method for the elimination of Fe ions from a buffer solution is described. To 2000 ml of a buffer solution containing 25% of sodium acetate and 10% sodium citrate having pH limits of 4.0 - 5.5, 20 ml of a 1% solution of 1,10-phenanthroline and 2g of solid sodium perchlorate are added. The solution is irradiated for 1 hour by a 200W mercury lamp, and the solution is extracted by 50ml of chloroform; the extraction is repeated until the sample is colorless. Traces of chloroform are removed from the solution by filtration over a filter.
[JPRS: 36,002]

SUB CODE: 07 / SUBM DATE: 07May65 / ORIG REF: 008 / Sov REF: 001
OTH REF: 009

Card 1/1 bba,

091P 0751

I 42262-66 E&P(j) WN/JW/RM
ACC NR: AP6031478

SOURCE CODE: CZ/0008/66/000/003/0385/0430

45
B

AUTHOR: Novak, Josef P.

ORG: Department of Physical Chemistry, College of Chemical Technology, Prague
(Katedra fysikalni chemie, Vysoka skola chemicko-technologicka)

TITLE: Calculation of thermodynamic properties of real gas systems

SOURCE: Chemicke listy, no. 3, 1966, 385-430

TOPIC TAGS: enthalpy, entropy, real gas

ABSTRACT: General expressions for the calculation of thermodynamic factors of gases and their mixtures are discussed. These functions were also related to conditions existing on the basis of the equilibrium conditions (such as those determined from Van der Waal's equation). Internal energy, enthalpy, entropy, free energy, free enthalpy, molar heats, and fugacity of gases are discussed. Thermodynamic factors in multicomponent systems are reviewed. Determination of partial molar values, and the state behavior of mixtures are described. The laws of Dalton, and Amagat, the rule of Bartlett, and that of Kricevsky are discussed. Comparison of various methods of calculation is presented. Orig. art. has: 2 formulas. [JPRS: 36,002]

SUB CODE: 20 / SUBM DATE: 26Oct64 / SOV REF: 001 / OTH REF: 047

Card 1/1 bkh

0918 0758

Novak, J.

FISCHER, V;NOVAK, J.

Experiences with undecylenic acid. Cesk. derm. 25 no.7-8:
273-278 July 1950. (CML 20:1)

1. Of the First Dermato-Venereological Clinic in Prague (Head—
Prof. K. Gawalowski, M. D.).

NOVAK J.

NOVAK J.

Koreni hladiny penicilliu v krvi. [Determination of penicillin in blood]
Cesk. dom. 25:4 1 Apr 50 p. 147-50.

1. Of the First Skin Clinic in Prague (Head--K. Cawalowski, M.D.).

CIVIL 19, 5, Nov. 50

NOVAK, J.

Surgical sterilization. Prakt. lek., Praha 32 no.3:57-61 5 Feb
1952. (CLML 22:2)

1. Of the First Dermato-Venereological Clinic (Head--Prof.
Gawalowski, M. D.) in Prague.

NOVAK, J.

Treatment of infectious eczema. Cesk. derm. 27 no.3-4:150-153
June 1952. (CLME 22:3)

l. Of the First Dermatological Clinic (Head--Prof. K. Gawalow-
ski, M. D.) of Charles University, Prague.

NOVAK, J.

GAWALOWSKI, K; NOVAK, J.; PROCHAZKA, K. "Indication of Penicillin Therapy in Dermatology." p. 194.
(Casopis Lekaru Ceskych. Vol. 93, no. 8, Feb. 1954. Praha).

SO: Monthly List of Acquisitions, East European Accessions, Vol. 3, No. 6, Library of Congress, June 1958, Unclassified.